

A stereoselective synthesis of verbalactone—determination of absolute stereochemistry[☆]

G. V. M. Sharma* and Ch. Govardhan Reddy

D-211, Discovery Laboratory, Organic Chemistry Division-III, Indian Institute of Chemical Technology, Hyderabad 500 007, India

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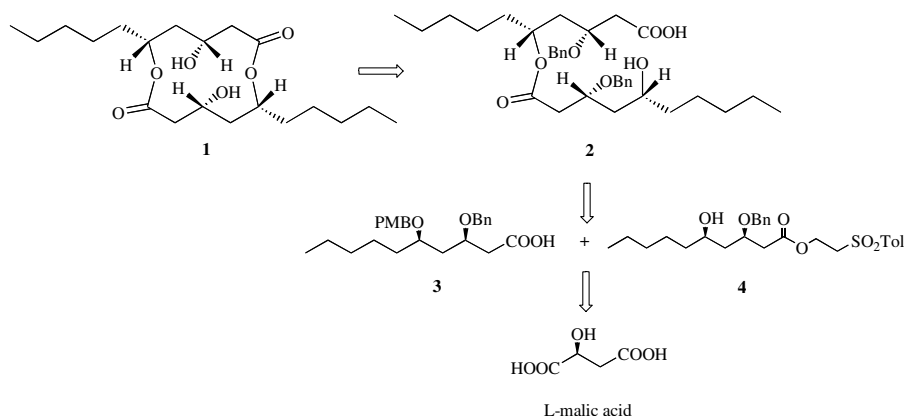
Abstract—A total synthesis of verbalactone has been achieved starting from L-malic acid. The two appropriately protected acid and alcohol segments were prepared from L-malic acid and lactonized under Yamaguchi conditions.

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Verbalactone (**1**), a novel macrocyclic dimer lactone, was isolated from the roots of *Verbascum undulatum* and exhibited interesting antibacterial activity.¹ It was the first example where a 1,7-dioxacyclododecane moiety was reported as the ring system of a natural product. The structure and the absolute stereochemistry of **1** as 4*R*,6*R*,10*R*,12*R* was determined by spectral methods (1D and 2D NMR, MS) and chemical correlation. Verbalactone (**1**) is thus a dimeric lactone with C₂-symmetry and has a NMR profile similar to the monomer lactone of (3*R*,5*R*)-dihydroxydecanoic acid.² In continuation of our interest on the synthesis of natural lactones,³ we

herein report a stereoselective synthesis of verbalactone (**1**).

The retrosynthetic plan for **1** (Scheme 1) indicated that the *seco* acid **2** could be a late stage intermediate. Acid **2**, in turn, could be prepared by coupling **3** and **4** via an esterification, while **3** and **4** could be prepared from L-malic acid. Thus, the basic strategy for the synthesis of the key fragment **3** was to utilize the lone stereocenter in malic acid as C-5 and to introduce the C-3 hydroxyl of the target molecule by an asymmetric Sharpless epoxidation.



Scheme 1. Retrosynthetic analysis of verbalactone **1**.

Keywords: Verbalactone; Sharpless epoxidation; Macrolactonization; Absolute stereochemistry.

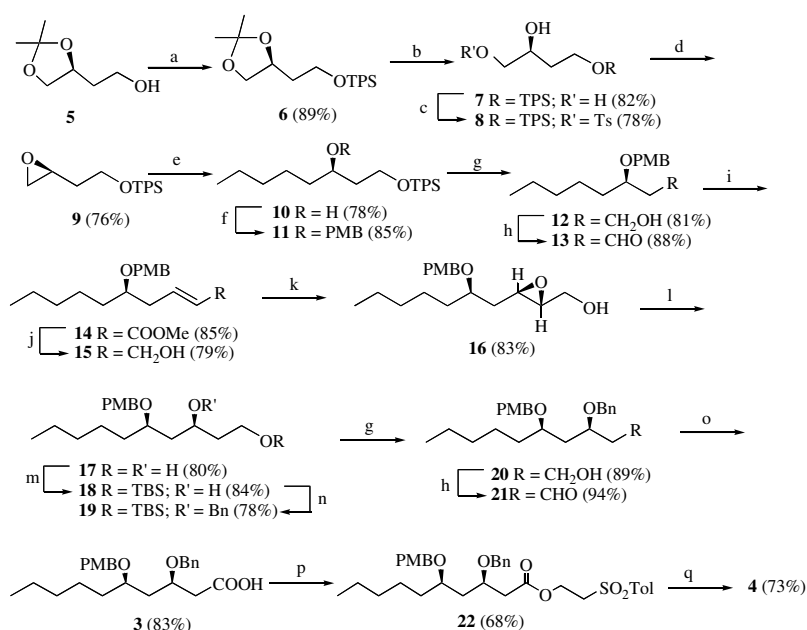
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* Corresponding author. Fax: +91 40 27160387; e-mail: esmvee@iict.res.in

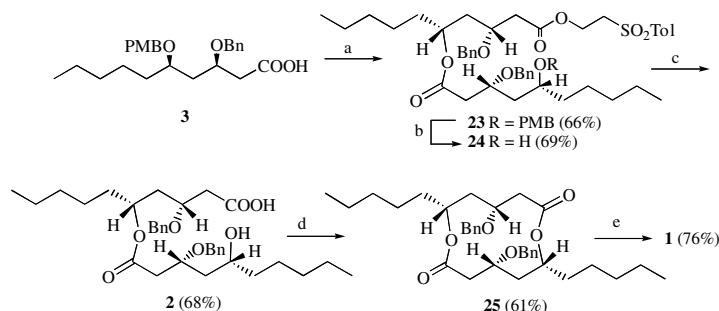
Accordingly, the known alcohol **5**⁴ (Scheme 2), derived from L-malic acid, was silylated (TBDPSCl, imidazole, CH₂Cl₂) to give **6** (89%), which on hydrolysis (60% aq AcOH) gave the diol **7** (82%). Selective tosylation of **7** with *p*-TsCl and Et₃N in CH₂Cl₂ furnished **8** (78%), which was reacted with 2 equiv of K₂CO₃ in MeOH to give the epoxide **9** (76%). Selective opening of **9** with CuI and *n*-BuLi⁵ furnished **10** in 78% yield (Scheme 2), which on protection with PMBBr gave **11** in 85% yield. Desilylation of **11** with TBAF gave alcohol **12** (81%), which on oxidation under Swern conditions and homologation of **13** with (methoxycarbonylmethylene)triphenyl phosphorane in benzene at reflux afforded ester **14** in 85% yield. Reduction of **14** using DIBAL-H furnished the alcohol **15** (79%), which on Sharpless asymmetric epoxidation⁶ with cumene hydroperoxide, (–)-DIPT and Ti(O*i*-Pr)₄ exclusively gave the epoxide **16** in 83% yield. Regioselective reductive opening of **16**

with NaAlH₂(OCH₂CH₂OMe)₂⁷ gave the 1,3-diol **17** in 80% yield. Silylation (TBDMSCl, imidazole, CH₂Cl₂) of **17** and subsequent benzylation of the secondary alcohol in **18** using NaH and BnBr gave **19** (78%). Deprotection of the TBDMS ether in **19** with TBAF and subsequent oxidation of **20** under Swern conditions furnished the corresponding aldehyde **21** (94%), which on further oxidation using NaClO₂⁸ and 30% H₂O₂ in *t*-BuOH/H₂O (2:1) gave **3** in 83% yield.

Having completed a synthesis of the key fragment **3**, it was subjected to esterification with *p*-toluenesulfonyl-ethanol via the mixed anhydride prepared on reaction of **3** with 2,4,6-trichlorobenzoyl chloride (Et₃N, THF) in the presence of DMAP to afford ester **22** (68%). Since the cleavage of *p*-toluenesulfonyl-ethyl groups with DBN^{3,9} is very facile and highly selective and does not effect other ester groups, it was thus chosen for the



Scheme 2. Reagents and conditions: (a) TBDPSCl, imidazole, CH₂Cl₂, rt, 5 h; (b) 60% aq AcOH, rt, 12 h; (c) *p*-TsCl, Et₃N, CH₂Cl₂, 0 °C, 10 h; (d) K₂CO₃, MeOH, rt, 2 h; (e) CuI, *n*-BuLi, dry ether, 0 °C, 2 h; (f) PMBBr, NaH, THF, 0 °C—rt, 5 h; (g) TBAF, CH₂Cl₂, rt, 10 h; (h) (COCl)₂ DMSO, Et₃N, CH₂Cl₂, –78 °C, 3 h; (i) Ph₃P=CHCOOMe, benzene, reflux, 2 h; (j) DIBAL-H, CH₂Cl₂, rt, 3 h; (k) (–)-DIPT, Ti(O*i*-Pr)₄ cumene hydroperoxide, 4 Å—MS, CH₂Cl₂, –20 °C, 3 h; (l) NaAlH₂(OCH₂CH₂OMe)₂, THF, 0 °C, 5 h; (m) TBDMSCl, imidazole, CH₂Cl₂, rt, 6 h; (n) BnBr, NaH, THF, 0 °C—rt, 4 h; (o) NaClO₂, 30% H₂O₂, *t*-BuOH/H₂O (2:1), 0 °C—rt, 10 h; (p) 2,4,6-trichlorobenzoyl chloride, Et₃N, THF, HOCH₂CH₂SO₂Ar, DMAP, toluene, rt, 16 h; (q) DDQ, CH₂Cl₂/H₂O (19:1), 0 °C—rt, 5 h.



Scheme 3. Reagents and conditions: (a) 2,4,6-trichlorobenzoyl chloride, Et₃N, THF, **4**, DMAP, toluene, rt, 20 h; (b) DDQ, CH₂Cl₂/H₂O (19:1), rt, 3 h; (c) DBN, benzene, rt, 12 h; (d) 2,4,6-trichlorobenzoyl chloride, Et₃N, THF, DMAP, toluene, 90 °C, 24 h; (e) TiCl₄, CH₂Cl₂, 0 °C, 1 h.

protection of acid **3**, in the present synthesis. Oxidative deprotection of the PMB group in **22** using DDQ in aq CH₂Cl₂ gave the hydroxy ester **4** in 73% yield, ready for coupling with segment **3**.

Esterification of acid **3** with alcohol **4** under Yamaguchi conditions gave **23** (66%), which on PMB deprotection furnished **24** in 69% yield (Scheme 3). Selective cleavage of the *p*-toluenesulfonylethyl group in **24** was effected with DBN in C₆H₆ to give seco acid **2** (68%), which on lactonization under Yamaguchi conditions¹⁰ (2,4,6-trichlorobenzoyl chloride, Et₃N, THF, DMAP, toluene) afforded **25** in 61% yield. Finally, deprotection of the benzyl groups in **24** was effected with TiCl₄ (1 equiv) in CH₂Cl₂ at 0 °C to give synthetic **1** as a colorless oil in 76% yield, $[\alpha]_D^{25} + 8.2$ (*c* 0.3, CHCl₃). The ¹H and ¹³C NMR data and optical rotation value of synthetic **1**¹¹ were in good accordance with those of the natural product.¹

In conclusion, a synthesis of verbalactone (**1**) by a combination of chiral pool and asymmetric approaches has been achieved.

Acknowledgements

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- Spectral data of verbalactone (**1**): Colorless oil, $[\alpha]_D^{25} + 8.2$ (*c* 0.3, CHCl₃); lit.¹ $[\alpha]_D^{25} + 7.3$ (*c* 0.9, CHCl₃); IR (neat): 3518, 1710, 1265, 1170 cm⁻¹. ¹H NMR (CDCl₃, 400 MHz): δ 0.85 (t, 6H, *J* = 7.0 Hz, H-17, 22), 1.22–1.31 (m, 12H, H-14, 15, 16, 19, 20, 21), 1.49–1.55 (m, 4H, H-13, 18), 1.96 (td, 2H, *J* = 15.2, 4.2 Hz, H-5a, 11a), 2.05 (ddd, 2H, *J* = 15.2, 10.2, 3.1 Hz, H-5b, 11b), 2.68 (d, 4H, *J* = 3.6 Hz, H-3, 9), 3.73 (br s, 2H, 3-OH, 10-OH), 4.06 (ddd, 2H, *J* = 4.2, 3.6, 3.1 Hz, H-4, 10), 4.94 (ddd, 2H, *J* = 10.2, 4.7, 4.6 Hz, H-6, 12); ¹³C NMR (CDCl₃, 300 MHz): δ 13.91, 22.45, 25.52, 31.42, 31.78, 38.24, 39.29, 64.97, 72.57, 172.91; FAB MS (*m/z*, %): 373 (M⁺+1, 32), 187 (94), 151 (26), 127 (100), 93 (34), 55 (44).